

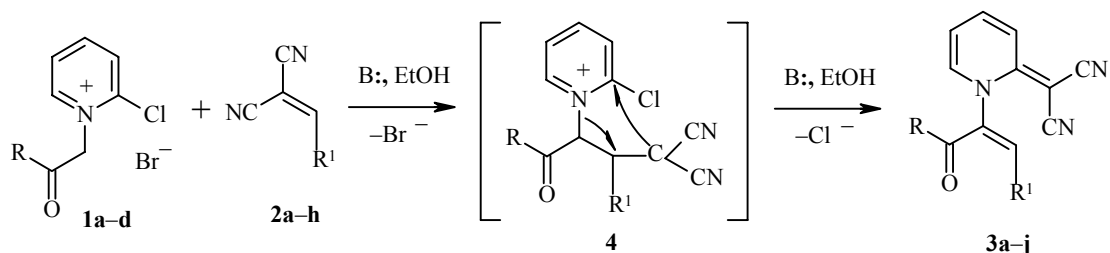
**SYNTHESIS OF 1-(1-AROYL-  
2-ARYLVINYL)-2-DICYANOMETHYLEN-  
1,2-DIHYDROPYRIDINES FROM  
2-CHLOROPYRIDINIUM SALTS  
AND UNSATURATED NITRILES**

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*1-(1-Aroyl-2-arylvinyl)-2-dicyanomethylene-1,2-dihydropyridines are formed from 1-(aroylmethyl)-2-chloropyridinium bromides and arylmethylenemalonitriles in ethanol at room temperature in the presence of a twofold excess of triethylamine. The products are converted into 2-amino-3-aryol-1-cyanoindolizines on boiling in acetic acid.*

**Keywords:** 1-(aroylmethyl)-2-chloropyridinium bromides, reactions with arylmethylenemalonitriles, 1-(1-aryol-2-arylvinyl)-2-dicyanomethylene-1,2-dihydropyridines.

In a continuation of investigations, which are aimed at simple methods for the synthesis of 2-dicyanomethylenepyridines [1, 2], it was found that 1-(aroylmethyl)-2-chloropyridinium bromides **1a-d** reacted with arylmethylenemalononitriles **2a-h** in the presence of a twofold excess of a tertiary base to give 1-(1-aryol-2-arylvinyl)-2-dicyanomethylene-1,2-dihydropyridines **3a-j** [3] (Table 1). The reactions occur under mild conditions, probably via intermediate **4**. The structures of compounds **3a-j** have been confirmed by IR and <sup>1</sup>H NMR spectroscopy (Table 2).



**1 a** R = C<sub>6</sub>H<sub>5</sub>, **b** R = 4-BrC<sub>6</sub>H<sub>4</sub>, **c** R = 4-ClC<sub>6</sub>H<sub>4</sub>, **d** 2-thienyl; **2 a** R<sup>1</sup> = 4-CH<sub>3</sub>OC<sub>6</sub>H<sub>4</sub>,  
**b** R<sup>1</sup> = 3-FC<sub>6</sub>H<sub>4</sub>, **c** R<sup>1</sup> = 4-FC<sub>6</sub>H<sub>4</sub>, **d** R<sup>1</sup> = 3,4-(CH<sub>3</sub>O)<sub>2</sub>C<sub>6</sub>H<sub>3</sub>, **e** R<sup>1</sup> = 2,4-(C<sub>2</sub>H<sub>5</sub>O)<sub>2</sub>C<sub>6</sub>H<sub>3</sub>,  
**f** R<sup>1</sup> = 2-ClC<sub>6</sub>H<sub>4</sub>, **g** R<sup>1</sup> = 2-furyl, **h** R<sup>1</sup> = C<sub>6</sub>H<sub>5</sub>; **3 a** R = C<sub>6</sub>H<sub>5</sub>, R<sup>1</sup> = 4-CH<sub>3</sub>OC<sub>6</sub>H<sub>4</sub>, **b** R = C<sub>6</sub>H<sub>5</sub>,  
R<sup>1</sup> = 3-FC<sub>6</sub>H<sub>4</sub>, **c** R = C<sub>6</sub>H<sub>5</sub>, R<sup>1</sup> = 4-FC<sub>6</sub>H<sub>4</sub>, **d** R = C<sub>6</sub>H<sub>5</sub>, R<sup>1</sup> = 3,4-(CH<sub>3</sub>O)<sub>2</sub>C<sub>6</sub>H<sub>3</sub>, **e** R = C<sub>6</sub>H<sub>5</sub>,  
R<sup>1</sup> = 2,4-(C<sub>2</sub>H<sub>5</sub>O)<sub>2</sub>C<sub>6</sub>H<sub>3</sub>, **f** R = C<sub>6</sub>H<sub>5</sub>, R<sup>1</sup> = 2-ClC<sub>6</sub>H<sub>4</sub>, **g** R = 4-BrC<sub>6</sub>H<sub>4</sub>, R<sup>1</sup> = 4-FC<sub>6</sub>H<sub>4</sub>,  
**h** R = 4-BrC<sub>6</sub>H<sub>4</sub>, R<sup>1</sup> = 2-furyl, **i** R = 4-ClC<sub>6</sub>H<sub>4</sub>, R<sup>1</sup> = 2-furyl, **j** R = 2-thienyl, R<sup>1</sup> = C<sub>6</sub>H<sub>5</sub>; B = Et<sub>3</sub>N

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TABLE 1. Characteristics of Compounds **3a-j**

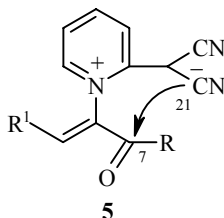
Compound	Empirical formula	Found, %			mp, °C (recrystallisation solvent)	Yield, %
		Calculated, %				
		C	H	N		
<b>3a</b>	C <sub>24</sub> H <sub>17</sub> N <sub>3</sub> O <sub>2</sub>	76.01	4.48	11.11	146-148 (EtOH)	88
		75.97	4.52	11.08		
<b>3b</b>	C <sub>23</sub> H <sub>14</sub> FN <sub>3</sub> O	75.22	3.87	11.41	184-186 (EtOH)	87
		75.19	3.84	11.44		
<b>3c</b>	C <sub>23</sub> H <sub>14</sub> FN <sub>3</sub> O	75.23	3.82	11.46	126-128 (EtOH)	89
		75.19	3.84	11.44		
<b>3d</b>	C <sub>25</sub> H <sub>19</sub> N <sub>3</sub> O <sub>3</sub>	73.31	4.71	10.29	222-224 (EtOH)	73
		73.34	4.68	10.26		
<b>3e</b>	C <sub>27</sub> H <sub>23</sub> N <sub>3</sub> O <sub>3</sub>	74.09	5.31	9.64	150-152 (BuOH)	74
		74.12	5.30	9.60		
<b>3f</b>	C <sub>23</sub> H <sub>14</sub> ClN <sub>3</sub> O	71.93	3.63	10.99	189-191 (EtOH)	59
		71.97	3.68	10.95		
<b>3g</b>	C <sub>23</sub> H <sub>13</sub> BrFN <sub>3</sub> O	61.94	2.98	9.39	208-209 (BuOH)	64
		61.90	2.94	9.42		
<b>3h</b>	C <sub>21</sub> H <sub>12</sub> BrN <sub>3</sub> O <sub>2</sub>	60.35	2.93	10.08	240-242	83
		60.31	2.89	10.05		
<b>3i</b>	C <sub>21</sub> H <sub>12</sub> ClN <sub>3</sub> O <sub>2</sub>	67.52	3.21	11.27	261-262 (dec.)	77
		67.48	3.24	11.24		
<b>3j</b>	C <sub>21</sub> H <sub>13</sub> N <sub>3</sub> OS	71.01	3.72	11.79	138-140 (EtOH)	82
		70.97	3.69	11.82		

In addition the molecular structure of 1-[1-benzoyl-2-(2-furyl)vinyl]-2-dicyanomethylene-1,2-dihydropyridine was investigated by X-ray crystallography [4].

TABLE 2. IR and <sup>1</sup>H NMR Spectroscopic Data for Compounds **3a-j**

Compound	IR spectrum, ν, cm <sup>-1</sup>			<sup>1</sup> H NMR spectrum, δ, ppm
	C=N	C=O	C=C	
<b>3a</b>	2157, 2180	1633	1620	3.80 (3H, s, OCH <sub>3</sub> ); 6.94 (1H, s, 4-H); 7.02-8.18 (13H, m, 3-, 5-, 6-H, C <sub>6</sub> H <sub>4</sub> , C <sub>6</sub> H <sub>5</sub> , CH)
<b>3b</b>	2155, 2186	1653	1624	6.85-8.07 (13H, m, Py, C <sub>6</sub> H <sub>5</sub> , C <sub>6</sub> H <sub>4</sub> ); 8.20 (1H, s, CH)
<b>3c</b>	2154, 2184	1649	1621	6.96 (1H, t, 4-H); 7.13-8.10 (13H, m, 3-, 5-, 6-H, C <sub>6</sub> H <sub>4</sub> , C <sub>6</sub> H <sub>5</sub> ); 17 (1H, s, CH)
<b>3d</b>	2162, 2186	1652	1623	3.58 (3H, s, OCH <sub>3</sub> ); 3.81 (3H, s, OCH <sub>3</sub> ); 6.71-8.05 (12H, m, Py, C <sub>6</sub> H <sub>5</sub> , C <sub>6</sub> H <sub>3</sub> ); 7.79 (1H, s, CH)
<b>3e</b>	2160, 2168 2185, 2195	1650	1622	1.26 (6H, t, (CH <sub>3</sub> ) <sub>2</sub> ); 4.08 (4H, dd, (OCH <sub>2</sub> ) <sub>2</sub> ); 6.64 (2H, d, C <sub>6</sub> H <sub>2</sub> ); 6.98 (1H, t, 4-H); 7.33-7.98 (9H, m, 3-, 5-, 6-H, C <sub>6</sub> H <sub>1</sub> , C <sub>6</sub> H <sub>5</sub> ); 8.25 (1H, s, CH)
<b>3f</b>	2152, 2180	1656	1620	6.90 (1H, t, 4-H); 6.93-8.12 (13H, m, 3-, 5-, 6-H, C <sub>6</sub> H <sub>4</sub> , C <sub>6</sub> H <sub>5</sub> , CH)
<b>3g</b>	2154, 2182	1654	1622	6.98 (1H, t, 4-H); 7.31-7.98 (12H, m, 3-, 5-, 6-H, C <sub>6</sub> H <sub>4</sub> , C <sub>6</sub> H <sub>4</sub> , CH)
<b>3h</b>	2158, 2187	1651	1624	6.74 (1H, dd, C <sub>4</sub> H <sub>3</sub> O); 6.92 (1H, t, 4-H); 7.17 (1H, d, C <sub>4</sub> H <sub>3</sub> O); 7.35 (1H, d, C <sub>4</sub> H <sub>3</sub> O); 7.73-7.92 (8H, m, 3-, 5-, 6-H, C <sub>6</sub> H <sub>4</sub> , CH)
<b>3i</b>	2152, 2179	1648	1620	6.72 (1H, dd, C <sub>4</sub> H <sub>3</sub> O); 6.94 (1H, t, 4-H); 7.15 (1H, d, C <sub>4</sub> H <sub>3</sub> O); 7.35 (1H, d, C <sub>4</sub> H <sub>3</sub> O); 7.75-8.01 (7H, m, 3-, 5-, 6-H, C <sub>6</sub> H <sub>4</sub> ); 8.23 (1H, s, CH)
<b>3j</b>	2162, 2192	1656	1624	6.72-7.00 (4H, m, 4-H, C <sub>4</sub> H <sub>3</sub> S); 7.20-8.10 (8H, m, 5-, 6-, C <sub>6</sub> H <sub>5</sub> , CH); 8.20 (1H, d, 3-H)

The considerable decrease in the vibration frequencies of the cyano groups to 2152-2195  $\text{cm}^{-1}$  in the IR spectra and the increase intensities in comparison with 2(1H)-dicyanomethylenepyrimidines (quinolines) [1, 5] indicate that the bipolar resonance form **5** contributes to the structure of compounds **3**.



The decrease in the frequencies of the keto groups to 1633-1656  $\text{cm}^{-1}$  [6] indicates a large amount of conjugation in the  $\alpha,\beta$ -unsaturated ketone unit. In the  $^1\text{H}$  NMR spectra the olefinic proton is shifted to low field at 7.67-8.03 ppm which also confirmed conjugation of the double bond and the keto group. On the whole the  $^1\text{H}$  NMR spectra are not in contradictory with suggested structure **3**, however the basic part of signals were overlapped by the resonances of the aromatic protons at 6.71-8.21 ppm.

It was predicted with high probability on the basis of the X-ray crystallographic results (an enforced short nonbonding intramolecular contact C(7)···C(21) [2.797(4) Å] [4]) and spectroscopic studies (redistribution of the electron density in molecules **3**) that compounds **3** would undergo further cyclisation to the indolizines **6**. In fact when compounds **3c,g,i** were boiled for a short time in acetic acid they gave 2-amino-3-aroyl-1-cyanoindolizines **6a-c** (route A) (in small yields because of insufficient solubility). The  $^1\text{H}$  NMR spectra and characteristics of these compounds are given in Table 3.

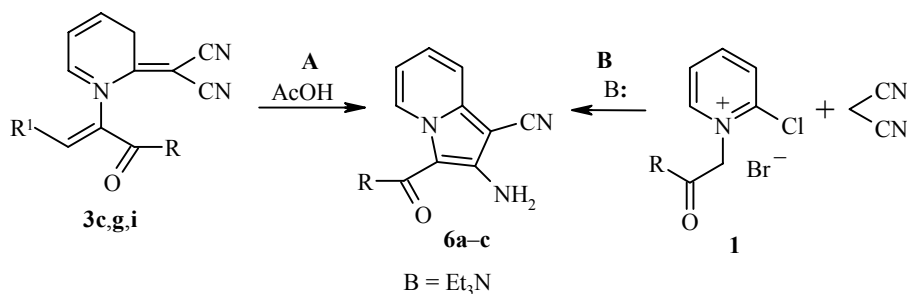


TABLE 3. Characteristics and  $^1\text{H}$  NMR Spectra of Compounds **6a-c**

Compound	Empirical formula	Found, %			mp, °C (ethanol)	$^1\text{H}$ NMR spectrum, $\delta$ , ppm	Yield, % A/B
		Calculated, %					
		C	H	N			
<b>6a</b>	$\text{C}_{16}\text{H}_{11}\text{N}_3\text{O}$	$\frac{73.58}{73.55}$	$\frac{4.22}{4.24}$	$\frac{16.04}{16.08}$	161-163	5.66 (2H, br. s, $\text{NH}_2$ ); 6.98 (1H, m, 7-H); 7.50 (2H, dd, 5-, 6-H); 7.58 (5H, s, $\text{C}_6\text{H}_5$ ); 9.16 (1H, d, 8-H)	32/90
<b>6b</b>	$\text{C}_{16}\text{H}_{10}\text{BrN}_3\text{O}$	$\frac{56.52}{56.49}$	$\frac{2.99}{2.96}$	$\frac{12.33}{12.35}$	192-194	5.92 (2H, br. s, $\text{NH}_2$ ); 7.00 (1H, m, 7-H); 7.55 (2H, dd, 5-, 6-H); 7.68 (4H, dd, $\text{C}_6\text{H}_4$ ); 9.17 (1H, d, 8-H)	43/94
<b>6c</b>	$\text{C}_{16}\text{H}_{10}\text{ClN}_3\text{O}$	$\frac{65.01}{64.98}$	$\frac{3.44}{3.41}$	$\frac{14.17}{14.21}$	199-201	5.88 (2H, br. s, $\text{NH}_2$ ); 6.97 (1H, dt, 7-H); 7.48 (2H, dd, 5-, 6-H); 7.60 (4H, s, $\text{C}_6\text{H}_4$ ); 9.14 (1H, d, 8-H)	41/96

The indolizines **6a-c** were also obtained by direct synthesis from the pyridinium salts **1a-c** and malonodinitrile (route **B**) [7].

## EXPERIMENTAL

IR spectra of nujol mulls were recorded with an IKS-29. <sup>1</sup>H NMR spectra were recorded with a Bruker WP-100 SY (100.13 MHz, internal standard TMS). Commercial DMSO-d<sub>6</sub> was used as solvent without further purification. Purity of the products was monitored by TLC (Silufol UV-254, 3:5 acetone–hexane).

**1-(1-Aroyl-2-arylviny)-2-dicyanomethylene-1,2-dihydropyridines (3a-j).** Triethylamine (0.7 ml, 5.0 mmol) was added with stirring to a suspension of a salt **1** (2.5 mmol) and an unsaturated nitrile **2** (2.5 mmol) in ethanol (15–20 ml). The mixture was stirred for 3 h at 20°C. The precipitate was filtered off and washed with ethanol and hexane. Where necessary it was recrystallized from a suitable solvent. Characteristics of the products are cited in Tables 1 and 2.

**2-Amino-3-aroyl-1-cyanoindolizines (6a-c).** **A.** A solution of a 1,2-dihydropyridine **3** (1.0 mmol) in acetic acid (5 ml) was boiled for 5 min and kept for 1 day at 20°C. The precipitate was filtered off and washed with ethanol and hexane.

**B.** Triethylamine (1.4 ml, 10.0 mmol) was added to a suspension of a salt **1** (5.0 mmol) and malononitrile (0.4 g, 6.0 mmol) in ethanol (15 ml). The mixture was stirred at 20°C for 1 h, then kept for 1 day in a refrigerator. The precipitate was filtered off, washed with ethanol and hexane, and recrystallized from ethanol. Characteristics of the compounds obtained are cited in Table 3.

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