

# Synthesis of 2-Thioxo-4,6-diphenyl-1,2-dihydronicotinonitrile via Condensation of Benzaldehyde with Cyanothioacetamide and *p*-(1-Styryl)morpholine

I. V. Dyachenko and V. D. Dyachenko

Shevchenko Lugansk University, ul. Oboronnaya 2, Lugansk, 91011 Ukraine  
e-mail: dyachvd@mail.ru

Received March 5, 2015

**Abstract**—2-Thioxo-4,6-diphenyl-1,2-dihydronicotinonitrile has been prepared via condensation of benzaldehyde with cyanothioacetamide and *N*-(1-styryl)morpholine; alkylation of the product with alkyl halides has afforded substituted 2-alkylsulfanyl-4,6-diphenylnicotinonitriles and 3-amino-2-acyl-4,6-diphenylthieno[2,3-*b*]pyridines.

**Keywords:** benzaldehyde, cyanothioacetamide, *N*-(1-styryl)morpholine, 2-thioxodihydronicotinonitrile, 2-alkylsulfanyllicotinonitrile, thieno[2,3-*b*]pyridine

**DOI:** 10.1134/S1070363215060146

Derivatives of aryl(hetaryl)-substituted 2-oxo(thioxo)-1,2-dihydronicotinonitrile are biologically active compounds exhibiting antimicrobial [1, 2], antitumor [3], and herbicide [4] activity. They can be prepared via interaction of  $\delta$ -ketodinitriles with elemental sulfur [5, 6] or hydrogen sulfide [7], chalcones reaction with cyanothioacetamide [7, 8], and by condensation of aryl(hetaryl)methylenecyanothioacetamides with acetophenones [9–11] or their enamines [11].

Extending the studies on the chemistry of 4,6-diaryl-2-thioxo-1,2-dihydronicotinic acid derivatives [12–14], we developed a new procedure for the preparation of such organic compounds based on the condensation of benzaldehyde (**I**) with cyanothioacetamide (**II**) and *N*-(1-styryl)morpholine (**III**) in anhydrous ethanol at 20°C; the reaction product was 2-thioxo-4,6-diphenyl-1,2-dihydronicotinonitrile (**IV**). The reaction likely involved the formation of Knoevenagel alkene (**A**) that alkylated the enamine **III** via the Stork reaction [15] to yield the adduct **B**. The latter underwent the intramolecular transamination [16] followed by dehydration (probably, aided by air oxygen) to form compound **IV**, earlier prepared via the reaction of  $\delta$ -ketodinitrile with elemental sulfur [5] (Scheme 1).

The alkylation of compound **IV** with alkyl halides **Va–Vm** in DMF solution in the alkali medium occurred regioselectively at the sulfur atom to yield the thioesters **VIa–VIm** capable of the intramolecular

cyclization at their treatment with KOH solution in aqueous DMF affording the 2-alkoxycarbonyl-3-amino-4,6-diphenylthieno[2,3-*b*]pyridines **VIIa–VIIc**, potential intermediate products for the synthesis of antitumor [17, 18] and antiviral [19] drugs.

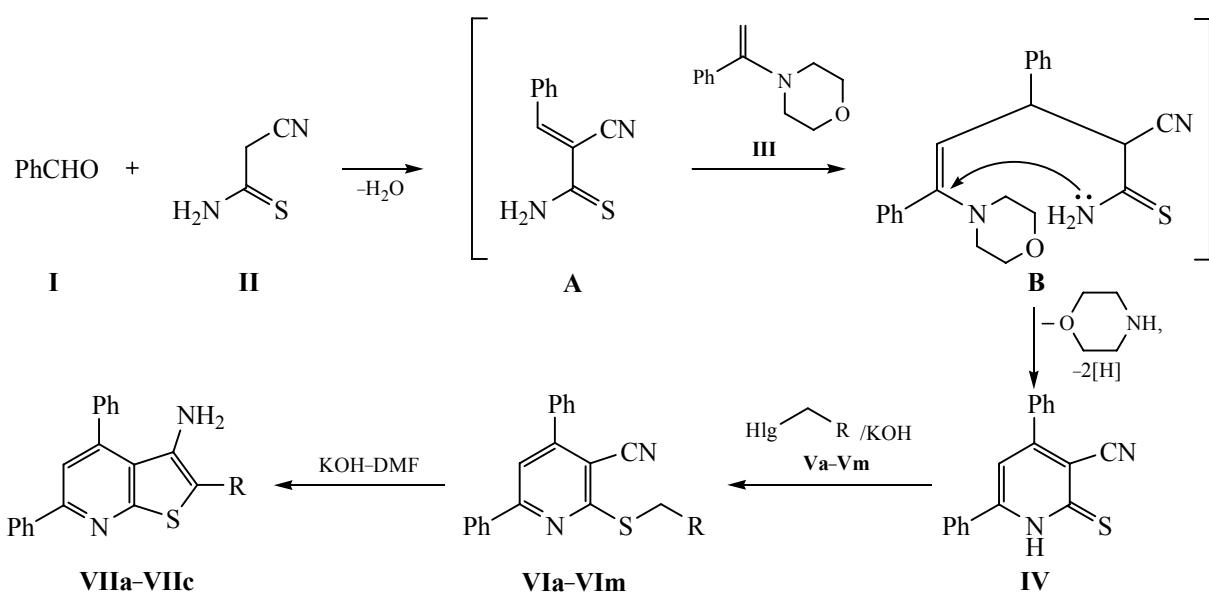
Structure of the prepared compounds **VIa–VIm** and **VIIa–VIIc** was confirmed by their physicochemical and spectral properties (Tables 1 and 2).

In particular, IR spectra of compounds **VI** contained characteristic absorption bands of the conjugated nitrile group stretching vibrations at 2214–2226 cm<sup>−1</sup>. Those signals were absent in the spectra of the thienopyridines **VII**; instead, the bands assigned to stretching and bending vibrations of the amino group were observed at 3198–3350 and 1638–1645 cm<sup>−1</sup>, respectively.

<sup>1</sup>H NMR spectra of the thioesters **VI** were remarkable for the presence of a singlet signal of the H<sup>5</sup> proton of pyrimidine ring at 7.38–7.85 ppm and that of the SCH<sub>2</sub> group at 4.11–5.11 ppm; those signals disappeared after the substrates transformation into the thienopyridines **VII**. <sup>1</sup>H NMR spectra of compounds **VII** contained the broadened singlet signal of the amino group protons at 5.80–6.51 ppm.

Mass spectra of compounds **VIb**, **VIi–VIm** contained weak signals of the [M + 2] ion peaks, pointing at the presence of a single S atom in the

Scheme 1.



**V–VII**,  $\text{Hlg} = \text{Br}$ ,  $\text{R} = \text{Me}_2\text{CHCH}_2\text{CO}$  (**a**);  $\text{Cl}$ ,  $\text{Me}_2\text{CHOCO}$  (**b**);  $\text{Br}$ ,  $\text{Me}(\text{CH}_2)_3\text{CO}$  (**c**);  $\text{Br}$ , 4-cyclohexylbenzoyl (**d**);  $\text{Br}$ ,  $\text{HC}\equiv\text{C}$  (**e**);  $\text{Br}$ ,  $\text{PhCO}$  (**f**);  $\text{Cl}$ ,  $\text{MeOCO}$  (**g**);  $\text{Cl}$ , quinolin-8-ylcarbamoyl (**h**);  $\text{Cl}$ ,  $\text{PhCH}_2\text{OCO}$  (**i**);  $\text{Cl}$ ,  $\text{PhNHCO}$  (**j**);  $\text{Cl}$ , 4-Ac $\text{C}_6\text{H}_4\text{NHCO}$  (**k**);  $\text{Cl}$ ,  $\text{Me}(\text{CH}_2)_7\text{OCO}$  (**l**);  $\text{Cl}$ ,  $\text{Me}(\text{CH}_2)_8\text{OCO}$  (**m**).

**Table 1.** Yields, melting points, and elemental analysis data for 2-alkylsulfanyl-4,6-diphenylnicotinonitriles (**VIa–VIIm**) and 3-amino-2-acyl-4,6-diphenylthieno[2,3-*b*]pyridines (**VIIa–VIIc**)

Comp. no.	Yield, %	mp, °C	Found, %			Formula	Calculated, %		
			C	H	N		C	H	N
<b>VIa</b>	82	105–106 (PrOH)	74.42	5.66	7.11	$\text{C}_{24}\text{H}_{22}\text{N}_2\text{OS}$	74.58	5.74	7.25
<b>VIb</b>	70	165–166 (AcOH)	71.01	5.04	7.12	$\text{C}_{23}\text{H}_{20}\text{N}_2\text{O}_2\text{S}$	71.11	5.19	7.21
<b>VIc</b>	74	110–112 (AcOH)	74.49	5.68	7.09	$\text{C}_{24}\text{H}_{22}\text{N}_2\text{OS}$	74.58	5.74	7.25
<b>VID</b>	85	176–178 (AcOH)	78.50	5.61	5.68	$\text{C}_{32}\text{H}_{28}\text{N}_2\text{OS}$	78.66	5.78	5.73
<b>VIe</b>	70	132–134 (EtOH)	77.14	4.25	8.41	$\text{C}_{21}\text{H}_{14}\text{N}_2\text{S}$	77.27	4.32	8.58
<b>VIIf</b>	81	182–184 (AcOH)	76.70	4.39	6.71	$\text{C}_{26}\text{H}_{18}\text{N}_2\text{OS}$	76.82	4.46	6.89
<b>VIg</b>	76	191–193 (MeOH)	69.83	4.32	7.60	$\text{C}_{21}\text{H}_{16}\text{N}_2\text{O}_2\text{S}$	69.98	4.48	7.77
<b>VIh</b>	77	211–213 (BuOH)	73.62	4.15	11.73	$\text{C}_{29}\text{H}_{20}\text{N}_4\text{OS}$	73.71	4.27	11.86
<b>VIi</b>	74	140–141 (EtOH)	74.08	4.50	6.32	$\text{C}_{27}\text{H}_{20}\text{N}_2\text{O}_2\text{S}$	74.29	4.62	6.42
<b>VIj</b>	82	218–220 (BuOH)	73.95	4.41	9.80	$\text{C}_{26}\text{H}_{19}\text{N}_3\text{OS}$	74.09	4.54	9.97
<b>VIk</b>	76	225–227 <sup>a</sup> (BuOH)	72.44	4.64	8.95	$\text{C}_{28}\text{H}_{21}\text{N}_3\text{O}_2\text{S}$	72.55	4.57	9.07
<b>VII</b>	75	98–99 (MeOH)	73.18	6.44	5.97	$\text{C}_{28}\text{H}_{30}\text{N}_2\text{O}_2\text{S}$	73.33	6.59	6.11
<b>VIIm</b>	69	96–97 (MeOH)	73.54	6.70	5.86	$\text{C}_{29}\text{H}_{32}\text{N}_2\text{O}_2\text{S}$	73.69	6.82	5.93
<b>VIIa</b>	79	100–102 (BuOH)	74.49	5.64	7.17	$\text{C}_{24}\text{H}_{22}\text{N}_2\text{OS}$	74.58	5.74	7.25
<b>VIIb</b>	68	115–116 ( <i>i</i> -PrOH)	70.96	5.10	7.12	$\text{C}_{23}\text{H}_{20}\text{N}_2\text{O}_2\text{S}$	71.11	5.19	7.21
<b>VIIc</b>	70	80–82 (MeOH)	74.40	5.64	7.15	$\text{C}_{24}\text{H}_{22}\text{N}_2\text{OS}$	74.58	5.74	7.25

<sup>a</sup> Sublimates at 160°C.

**Table 2.** IR and  $^1\text{H}$  NMR spectral data for 2-alkylsulfanyl-4,6-diphenylnicotinonitriles (**VIa–VIIm**) and 3-amino-2-acyl-4,6-diphenylthieno[2,3-*b*]pyridines (**VIIa–VIIc**)

Comp. no.	$\nu$ , $\text{cm}^{-1}$		$\delta$ , ppm ( $J$ , Hz)	
	C≡N or $\text{NH}_2$	C=O, $\delta(\text{NH}_2)$	H <sup>5</sup> , SCH <sub>2</sub> , or NH <sub>2</sub>	other signals
<b>VIa</b>	2218	1688	7.46, 4.18	1.91 d (6H, 2Me, $J$ 5.1), 2.03–2.15 m (1H, <u>CHMe<sub>2</sub></u> ), 2.54 d (2H, CH <sub>2</sub> CO, $J$ 6.0), 7.52–7.61 m (4H, Ph), 7.65–7.73 m (4H, Ph), 8.05 d (2H, Ph, $J$ 7.9)
<b>VIb</b>	2215	1719	7.85, 4.17	1.18 d (6H, 2Me, $J$ 4.8), 4.94 q (1H, <u>CHMe<sub>2</sub></u> , $J$ 4.8), 7.38–7.72 m (8H, H <sub>arom</sub> ), 8.11–8.26 m (2H, Ph)
<b>VIc</b>	2219	1713	7.38, 4.19	0.85 t (3H, Me, $J$ 7.5), 1.18–1.26 m (2H, CH <sub>2</sub> ), 1.43–1.59 m (2H, CH <sub>2</sub> ), 2.65 t (2H, CH <sub>2</sub> CO, $J$ 7.5), 7.44–7.58 m (6H, H <sub>arom</sub> ), 7.82–7.87 m (2H, Ph), 8.04 d (2H, Ph, $J$ 7.5)
<b>VID</b>	2220	1715	7.79, 4.92	1.21–1.62 m (5H, cyclohexane), 1.73–1.99 m (5H, cyclohexane), 2.68 t (1H, cyclohexane, $J$ 5.4), 7.21 t (2H, Ph, $J$ 6.9), 7.39 t (3H, Ph, $J$ 6.9), 7.55 t (3H, Ph, $J$ 7.1), 7.73 d (2H, Ph, $J$ 7.1), 7.90 d (2H, C <sub>6</sub> H <sub>4</sub> , $J$ 7.6), 8.05 d (2H, C <sub>6</sub> H <sub>4</sub> , $J$ 7.6)
<b>VIe</b>	2212	2244 (C≡C)	7.83, 4.21	2.90 s (1H, ≡CH), 7.48–7.63 m (6H, 2Ph), 7.71 d (2H, Ph, $J$ 6.9), 8.29 d (2H, Ph, $J$ 7.1)
<b>VIf</b>	2223	1700	7.69, 4.93	7.12 t (2H, Ph, $J$ 7.7), 7.33 t (1H, Ph, $J$ 6.4), 7.43–7.58 m (4H, Ph), 7.60–7.78 m (4H, Ph), 7.83 d (2H, Ph, $J$ 8.0), 8.09 d (2H, Ph, $J$ 8.0)
<b>VIg</b>	2218	1720	7.80, 4.18	3.78 s (3H, Me), 7.38–7.61 m (6H, 2Ph), 7.68 d (2H, Ph, $J$ 7.1), 8.14 d (2H, Ph, $J$ 6.9)
<b>VIh</b>	2221	1666	7.82, 4.44	7.16–7.65 m (10H, H <sub>arom</sub> ), 8.16–8.39 m (4H, H <sub>arom</sub> ), 8.51–8.72 m (2H, H <sub>arom</sub> ), 10.70 br.s (1H, NH)
<b>VIi</b>	2225	1714	7.76, 5.11	4.24 s (2H, OCH <sub>2</sub> ), 7.26 br.s (5H, Ph), 7.31–7.64 m (8H, Ph), 8.13 d (2H, Ph, $J$ 7.1)
<b>VIj</b>	2220	1670	7.72, 4.24	7.00 t (1H, Ph, $J$ 7.1), 7.09–7.33 m (5H, Ph), 7.48 br.s (5H, Ph), 7.64 d (2H, Ph, $J$ 6.9), 8.19 d (2H, Ph, $J$ 6.9), 10.23 br.s (1H, NH)
<b>VIk</b>	2222	1663	7.75 <sup>a</sup> , 4.29	2.54 s (3H, Me), 7.26–7.51 m (3H, H <sub>arom</sub> ), 7.57–7.90 <sup>a</sup> m (8H, H <sub>arom</sub> ), 7.96 t (2H, H <sub>arom</sub> , $J$ 7.1), 8.13 d (2H, H <sub>arom</sub> , $J$ 6.9), 10.60 br.s (1H, NH)
<b>VII</b>	2226	1718	7.74, 4.11	0.82 t (3H, Me, $J$ 6.7), 0.92–1.31 br.s (10H, 5CH <sub>2</sub> ), 1.39–1.64 m (2H, CH <sub>2</sub> ), 4.01 t (2H, OCH <sub>2</sub> , $J$ 6.8), 7.31–7.66 m (8H, H <sub>arom</sub> ), 8.02–8.19 m (2H, Ph)
<b>VIIm</b>	2225	1715	7.73, 4.13	0.85 t (3H, Me, $J$ 5.6), 1.01–1.39 br.s (12H, 6CH <sub>2</sub> ), 1.43–1.70 m (2H, CH <sub>2</sub> ), 4.06 t (2H, OCH <sub>2</sub> , $J$ 6.2), 7.41–7.67 m (8H, H <sub>arom</sub> ), 8.02–8.20 m (2H, H <sub>arom</sub> )
<b>VIIa</b>	3198, 3270, 3331	1690, 1638	7.78, 6.45	0.94 d (6H, 2Me, $J$ 5.2), 2.11–2.22 m (1H, <u>CHMe<sub>2</sub></u> ), 2.54 d (2H, CH <sub>2</sub> CO, $J$ 5.1), 7.41–7.46 m (2H, Ph), 7.49–7.66 m (6H, H <sub>arom</sub> ), 8.22 d (2H, Ph, $J$ 8.2)
<b>VIIb</b>	3205, 3272, 3345	1699, 1645	7.80, 5.80	1.30 d (6H, 2Me, $J$ 5.1), 5.06–5.19 m (1H, <u>CHMe<sub>2</sub></u> ), 7.48–7.53 m (3H, Ph), 7.62 br.s (5H, Ph), 8.23 d (2H, Ph, $J$ 7.1)
<b>VIIc</b>	3200, 3291, 3350	1710, 1644	7.76, 6.51	0.88 t (3H, Me, $J$ 5.6), 1.22–1.38 m (2H, CH <sub>2</sub> ), 1.59–1.63 m (2H, CH <sub>2</sub> ), 2.64 t (2H, OCH <sub>2</sub> , $J$ 5.5), 7.48 d (3H, Ph, $J$ 7.5), 7.58 br.s (5H, Ph), 8.21 d (2H, Ph, $J$ 7.6)

<sup>a</sup> Overlapping signals.

structures. The values of the molecular ion peaks coincided with the “nitrogen rule” [20].

To summarize, we have elaborated a new method of preparation of 2-thioxo-4,6-diphenyl-1,2-dihydronicotinonitriles based on the three-component condensation of benzaldehyde, cyanothioacetamide, and *N*-(1-styryl)morpholine.

## EXPERIMENTAL

IR spectra of the KBr pellets were recorded using an UR-20 instrument.  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectra of the solutions in  $\text{DMSO}-d_6$  were registered using a Bruker 500 spectrometer (at 500.13 and 125.75 MHz, respectively) with TMS as the internal reference. Elemental analysis was performed using an EuroVector EA-3000 microanalyzer. Chromato-mass spectrometry analysis was performed using a Hewlett-Packard 5890/5972 GC-MS spectrometer (chemical ionization, HP5-MS column, solutions in  $\text{CF}_3\text{COOH}$ ) for compounds **VIa**, **VIc–VIh** and **VIIa–VIIc** or a KRATOS 890A spectrometer (EI, 70 eV, direct injection of the substance in the ion source) for compounds **VIb**, **VIi–VIIm**. Melting points were determined using a Koeffler heating block. The reactions course was monitored by TLC on the Silufol-254 plates (acetone–hexane 3 : 5, developing with iodine vapor or UV irradiation).

**2-Thioxo-4,6-diphenyl-1,2-dihydronicotinonitrile (IV).** A droplet of morpholine was added to a mixture of 1.0 mL (10 mmol) of benzaldehyde and 1.0 g (10 mmol) of cyanothioacetamide, and the mixture was stirred during 30 min. Enamine **III** [1.9 g (10 mmol) at 20°C] was then added, and the mixture was further stirred during 1 h. Then, after maintaining for 24 h the mixture was diluted with 10 wt % aqueous hydrochloric acid and further kept at room temperature for 2 h. The formed precipitate was filtered off and washed with water, ethanol, and hexane. Yield 2.3 g (79%), yellow crystals, mp 227–229°C (AcOH) (mp 228–230°C [5]).

**2-Alkylsulfanyl-4,6-diphenylnicotinonitriles (VIa–VIIm) (general procedure).** 5.6 mL of 10 wt % aqueous KOH (10 mmol of the alkali) and 10 mmol of the corresponding alkyl halide **Va–Vm** were added to a stirred solution of 2.9 g (10 mmol) of compound **IV** in 15 mL of DMF, and the mixture was stirred during 30 min. After keeping for 24 h the mixture was diluted with equal volume of water; the formed precipitate was filtered off and washed with water, ethanol, and hexane.

**2-(2-Oxoisohexylsulfanyl)-4,6-diphenylnicotinonitrile (VIa).** Mass spectrum,  $m/z$  ( $I_{\text{rel}}$ , %): 387 (100)  $[M + 1]^+$ .  $M$  386.519. Fluorescent under UV irradiation.

**Isopropyl-2-(4,6-diphenyl-3-cyanopyridin-2-ylsulfanyl) acetate (VIb).** Mass spectrum,  $m/z$  ( $I_{\text{rel}}$ , %): 390 (3)  $[M + 2]^+$ , 389 (11)  $[M + 1]^+$ , 388 (33)  $[M]^+$ , 387 (39)  $[M - 1]^+$ , 345 (10), 329 (14), 301 (100)  $[M - \text{Me}_3\text{CHOCO}]^+$ , 255 (14), 227 (15), 140 (11), 77 (22)  $[\text{Ph}]^+$ , 44 (12)  $[\text{C}=\text{S}]^+$ , 43 (59)  $[\text{Me}_2\text{CH}]^+$ .  $M$  388.491.

**2-(2-Oxohexylsulfanyl)-4,6-diphenylnicotinonitrile (VIc).** Mass spectrum,  $m/z$  ( $I_{\text{rel}}$ , %): 387 (100)  $[M + 1]^+$ .  $M$  386.519. Fluorescent under UV irradiation.

**4,6-Diphenyl-2-[2-oxo-2-(4-cyclohexylphenyl)-ethylsulfanyl]nicotinonitrile (VID).** Mass spectrum,  $m/z$  ( $I_{\text{rel}}$ , %): 489 (100)  $[M + 1]^+$ .  $M$  488.656.

**2-Propargylsulfanyl-4,6-diphenylnicotinonitrile (VIe).** Mass spectrum,  $m/z$  ( $I_{\text{rel}}$ , %): 327 (100)  $[M + 1]^+$ .  $M$  326.423.

**2-Benzoylmethylsulfanyl-4,6-diphenylnicotinonitrile (VIf).** Mass spectrum,  $m/z$  ( $I_{\text{rel}}$ , %): 407 (100)  $[M + 1]^+$ .  $M$  406.509.

**Methyl-2-(4,6-diphenyl-3-cyanopyridin-2-yl) acetate (VIg).** Mass spectrum,  $m/z$  ( $I_{\text{rel}}$ , %): 361 (100)  $[M + 1]^+$ .  $M$  360.438.

**N-(Quinolin-8-yl)-2-(4,6-diphenyl-3-cyanopyridin-2-ylsulfanyl) acetamide (VIh).** Mass spectrum,  $m/z$  ( $I_{\text{rel}}$ , %): 472 (100)  $[M + 1]^+$ .  $M$  471.585.

**Benzyl-2-(4,6-diphenyl-3-cyanopyridin-2-ylsulfanyl) acetate (VIi).** Mass spectrum,  $m/z$  ( $I_{\text{rel}}$ , %): 438 (3)  $[M + 2]^+$ , 437 (9)  $[M + 1]^+$ , 436 (30)  $[M]^+$ , 435 (39)  $[M - 1]^+$ , 345 (5)  $[M - \text{PhCH}_2]^+$ , 301 (48), 288 (18), 255 (11), 227 (15), 91 (100)  $[\text{PhCH}_2]^+$ , 77 (21)  $[\text{Ph}]^+$ , 65 (12) 51 (5).  $M$  436.536.

**N-Phenyl-2-(4,6-diphenyl-3-cyanopyridin-2-ylsulfanyl) acetamide (VIj).** Mass spectrum,  $m/z$  ( $I_{\text{rel}}$ , %): 423 (4)  $[M + 2]^+$ , 422 (13)  $[M + 1]^+$ , 421 (45)  $[M]^+$ , 420 (5)  $[M - 1]^+$ , 329 (100)  $[M - \text{PhNH}]^+$ , 301 (56)  $[M - \text{PhNHCO}]^+$ , 255 (11), 211 (12), 93 (16)  $[\text{PhNH}_2]^+$ , 77 (15)  $[\text{Ph}]^+$ , 65 (7) 51 (5).  $M$  421.524.

**N-(4-Acetylphenyl)-2-(4,6-diphenyl-3-cyanopyridin-2-ylsulfanyl) acetamide (VIk).** Mass spectrum,  $m/z$  ( $I_{\text{rel}}$ , %): 465 (2)  $[M + 2]^+$ , 464 (4)  $[M + 1]^+$ , 463 (12)  $[M]^+$ , 462 (6)  $[M - 1]^+$ , 329 (35)  $[M - \text{AcC}_6\text{H}_4\text{NH}]^+$ , 301 (100)  $[M - \text{PhNH}]^+$ , 227 (13), 120 (14), 91 (7)  $[\text{PhN}]^+$ , 77 (16)  $[\text{Ph}]^+$ , 43 (15)  $[\text{Ac}]^+$ .  $M$  436.562.

**Octyl-2-(4,6-diphenyl-3-cyanopyridin-2-ylsulfanyl)acetate (VII).** Mass spectrum,  $m/z$  ( $I_{\text{rel}}$ , %): 460 (4)  $[M + 2]^+$ , 459 (19)  $[M + 1]^+$ , 458 (57)  $[M]^+$ , 457 (90)  $[M - 1]^+$ , 301 (100)  $[M - \text{Me}(\text{CH}_2)_7\text{OCO}]^+$ , 288 (11), 255 (15), 227 (13), 140 (6), 77 (16)  $[\text{Ph}]^+$ , 69 (9), 55 (18), 43 (47), 41 (39).  $M$  458.627.

**Nonyl-2-(4,6-diphenyl-3-cyanopyridin-2-ylsulfanyl)acetate (VIm).** Mass spectrum,  $m/z$  ( $I_{\text{rel}}$ , %): 474 (4)  $[M + 2]^+$ , 473 (16)  $[M + 1]^+$ , 472 (48)  $[M]^+$ , 471 (44)  $[M - 1]^+$ , 301 (100)  $[M - \text{Me}(\text{CH}_2)_8\text{OCO}]^+$ , 288 (9), 255 (12), 227 (11), 77 (8)  $[\text{Ph}]^+$ , 55 (17), 43 (39).  $M$  472.655.

**3-Amino-2-acyl-4,6-diphenylthieno[2,3-*b*]pyridines (VIIa–VIIc) (general procedure).** A mixture of 10 mmol of the corresponding 2-alkylsulfanylnicotinonitrile (VIa–VIc), 5.6 mL of 10 wt % aqueous KOH solution (10 mmol of the alkali), and 20 mL of DMF was stirred during 2 h and then diluted with equal volume of water. The formed precipitate was filtered off and washed with water, ethanol, and hexane.

**3-Amino-2-(1-oxoisooamyl)-4,6-diphenylthieno[2,3-*b*]pyridine (VIIa).** Mass spectrum,  $m/z$  ( $I_{\text{rel}}$ , %): 387 (100)  $[M + 1]^+$ .  $M$  386.519.

**Isopropyl-2-(3-amino-4,6-diphenylthieno[2,3-*b*]-pyridin-2-yl) acetate (VIIb).** Mass spectrum,  $m/z$  ( $I_{\text{rel}}$ , %): 389 (100)  $[M + 1]^+$ .  $^{13}\text{C}$  NMR spectrum,  $\delta_{\text{C}}$ , ppm: 22.28, 68.24, 95.98, 118.68, 120.84, 127.71, 129.11, 129.37, 129.78, 130.45, 136.82, 137.79, 147.76, 148.49, 156.80, 161.41, 164.45.  $M$  388.491.

**3-Amino-2-(1-oxoisooamyl)-4,6-diphenylthieno[2,3-*b*]-pyridine (VIIc).** Mass spectrum,  $m/z$  ( $I_{\text{rel}}$ , %): 387 (100)  $[M + 1]^+$ .  $M$  386.519. Fluorescent under UV irradiation.

## REFERENCES

1. Mohamed, S.F., Youssef, M.H., Amr, A.E.-G., and Kotb, E.R., *Scipharm.*, 2008, vol. 76, no. 2, p. 279. DOI: 10.3797/scipharm.0804-09.
2. Abdulla, M.M., *Monatsh. Chem.*, 2008, vol. 139, no. 6, p. 697. DOI: 10.1007/s00706-007-0804-1.
3. US Patent 6184237, 2001; *Ref. Zh. Khim.*, 2002, 02.04–19O.86P.
4. Germany Patent 10130357 2003; *Ref. Zh. Khim.*, 2003, 03.18–19O.347P.
5. Krauze, A.A., Bomika, Z.A., Shestopalov, A.M., Rodinovskaya, L.A., Pelcher, Yu.É., Dubur, G.Ya., Sharanin, Yu.A., and Promonenkov, V.K., *Chem. Heterocycl. Compd.*, 1981, vol. 17, no. 3, p. 279. DOI: 10.1007/BF00505994.
6. Shestopalov, A.M., Promonenkov, V.K., Sharanin, Yu.A., Rodinovskaya, L.A., and Sharanin, S.Yu., *Zh. Org. Khim.*, 1984, vol. 20, no. 7, p. 1517.
7. Krauze, A.A., Kalme, Z.A., Pelcher, Yu.É., Liepin'sh, É.É., Dipan, I.V., and Dubur, G.Ya., *Chem. Heterocycl. Compd.*, 1983, vol. 19, no. 11, p. 1202. DOI: 10.1007/BF00515357.
8. Abdel-Fattah, A.M., Elneairy, M.A.A., Gouda, M.N., and Attaby, F.A., *Phosphorus, Sulfur, Silicon, Relat. Elem.*, 2008, vol. 183, no. 7, p. 1592. DOI: 10.1080/10426500701693552.
9. Elgemeie, G.E.H., Elfohnan, H.A., and Nabey, H.A., *Sulfur Lett.*, 1989, vol. 9, nos. 1–2, p. 47.
10. Abdel-Rahman, A.E., Bakhite, E.A., Mohamed, O.S., and Thabet, E.A., *Phosphorus, Sulfur, Silicon, Relat. Elem.*, 2003, vol. 178, no. 1, p. 89. DOI: 10.1080/10426500307820.
11. Promonenkov, V.K., Shestopalov, A.M., Sharanin, Yu.A., Litvinov, V.P., Rodinovskaya, L.P., Zolotarev, B.M., Sokovykh, V.D., Rozynov, B.V., and Krymskii, Ya.Ya., *Zh. Org. Khim.*, 1985, vol. 21, no. 9, p. 1963.
12. Dyachenko, V.D., *Russ. J. Org. Chem.*, 2011, vol. 47, no. 10, p. 1535. DOI: 10.1134/S1070428011100150.
13. Dyachenko, V.D., *Chem. Heterocycl. Compd.*, 2005, vol. 41, no. 8, p. 1005. DOI: 10.1007/s10593-005-0268-3.
14. Dyachenko, V.D. and Chernega, A.N., *Chem. Heterocycl. Compd.*, 2005, vol. 41, no. 12, p. 1499. DOI: 10.1007/s10593-006-0027-0.
15. Stork, G., Brizzolara, A., Landesman, H., Szmuszkovich, J., and Terrell, R., *J. Am. Chem. Soc.*, 1963, vol. 85, no. 2, p. 207. DOI: 10.1021/ja00885a021.
16. Mishchenko, G.L. and Vatsuro, K.V., *Sinteticheskie metody organicheskoi khimii* (Synthetic Methods of Organic Chemistry), Moscow: Khimiya, 1982, p. 297.
17. EP Patent EP 1683799, 2006; *Ref. Zh. Khim.*, 2007, 07.20–19O.115P.
18. US Patent US 6964956, 2005; *Ref. Zh. Khim.*, 2006, 06.14–19O.69P.
19. EP Patent EP 1681292, 2006; *Ref. Zh. Khim.*, 2007, 07.20–19O.116P.
20. Pretsch, E., Biihlmann, P., and Affolter, C., *Structure Determination of Organic Compounds*, Berlin: Springer, 2000.