

# Novel Method of 4,6-Dimethyl-2-thioxo-1,2-dihydronicotinonitrile Synthesis

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**Abstract**—Condensation of acetaldehyde with cyanothioacetamide and 1-(prop-1-en-2-yl)pyrrolidine has afforded 4,6-dimethyl-2-thioxo-1,2-dihydronicotinonitrile. Alkylation of the latter with  $\alpha$ -haloketones yields substituted 2-alkylsulfanyl-4,6-dimethylnicotinonitriles and thieno[2,3-*b*]pyridines.

**Keywords:** acetaldehyde, cyanothioacetamide, 1-(prop-1-en-2-yl)pyrrolidine, 2-thioxo-1,2-dihydronicotinonitrile, 2-alkylsulfanylnicotinonitriles, thieno[2,3-*b*]pyridines

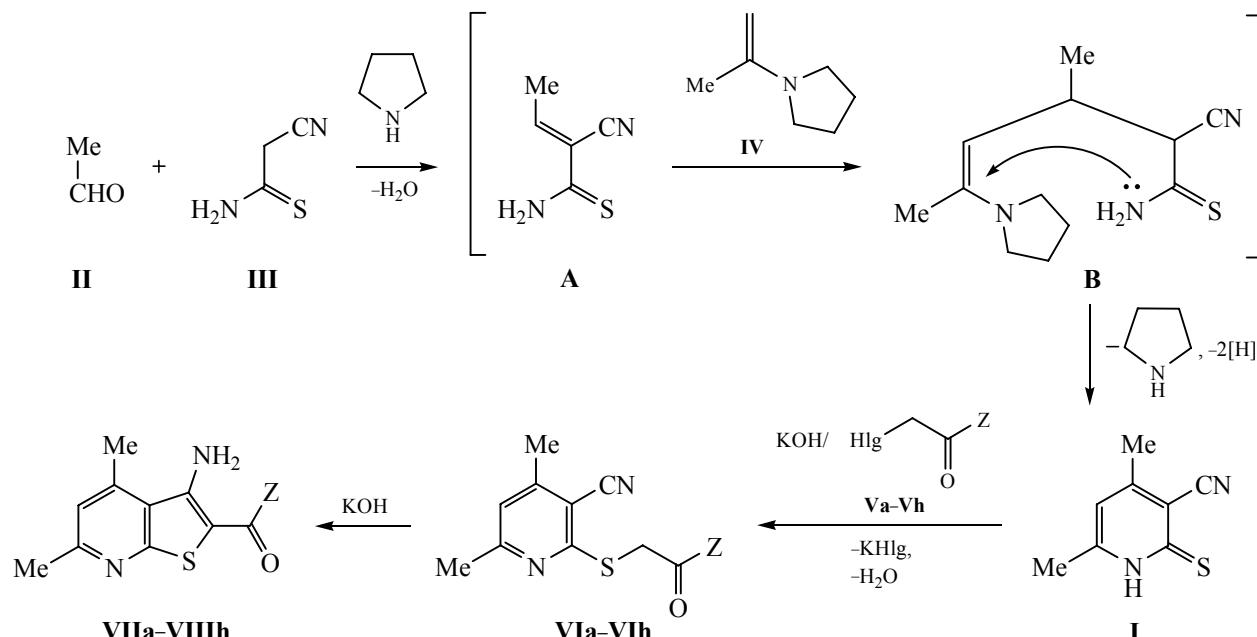
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Alkyl-substituted 2-thioxo(oxo)-1,2-dihydronicotinonitriles are intermediates in the synthesis of anti-inflammatory [1] and antineoplastic [2] drugs. They are components of medicines for Alzheimer's disease treatment [3, 4], herbicides [5], and azo dyes [6, 7]. The known methods of their synthesis are based on condensation of acetylacetone with cyanoacetamide or

cyanothioacetamide [8], nucleophilic substitution of the chlorine atom in 2-thiourechloronicotinonitrile [9], and the Michael reaction via the methylene components exchange [10].

In this work, we propose a new method of 4,6-dimethyl-2-thioxo-1,2-dihydronicotinonitrile **I** synthesis:

Scheme 1.



V–VII, Hlg = Br, Z = 3,4-Cl<sub>2</sub>C<sub>6</sub>H<sub>3</sub> (**a**); Br, 4-BuC<sub>6</sub>H<sub>4</sub> (**b**); Br, 4-PhC<sub>6</sub>H<sub>4</sub> (**c**); Cl, 4-BrC<sub>6</sub>H<sub>4</sub>NHCO (**d**); Br, 2,4-Me<sub>2</sub>C<sub>6</sub>H<sub>3</sub> (**e**); Br, 4-BrC<sub>6</sub>H<sub>4</sub> (**f**); Br, 3-BrC<sub>6</sub>H<sub>4</sub> (**g**); Cl, PhCH<sub>2</sub>O (**h**).

**Table 1.** Yields, melting points, and elemental analysis data for 2-alkylsulfanyl-4,6-dimethylnicotinonitriles **VIa–VIh** and 3-amino-4,6-dimethyl-2-Z-thieno[2,3-*b*]pyridines **VIIa–VIIh**

Comp. no.	Yield, %	mp, °C	Found, %			Formula	Calculated, %		
			C	H	N		C	H	N
<b>VIa</b>	89	118–120 (AcOH)	54.60	3.32	7.85	$C_{16}H_{12}Cl_2N_2OS$	54.71	3.44	7.98
<b>VIb</b>	75	80–81 (MeCN)	70.86	6.41	8.14	$C_{20}H_{22}N_2OS$	70.97	6.55	8.28
<b>VIc</b>	81	156–158 (BuOH)	73.69	4.92	7.77	$C_{22}H_{18}N_2OS$	73.72	5.06	7.81
<b>VID</b>	77	199–201 (AcOH)	50.95	3.66	11.02	$C_{16}H_{14}BrN_3OS$	51.07	3.75	11.17
<b>VIe</b>	70	116–118 (MeOH)	69.51	5.70	8.95	$C_{18}H_{18}N_2OS$	69.65	5.85	9.02
<b>VIIf</b>	82	125–127 (EtOH)	53.14	3.58	7.65	$C_{16}H_{13}BrN_2OS$	53.20	3.63	7.75
<b>VIg</b>	76	138–140 (AcOH)	53.11	3.55	7.61	$C_{16}H_{13}BrN_2OS$	53.20	3.63	7.75
<b>VIh</b>	73	81–83 (PrOH)	65.19	5.00	8.86	$C_{17}H_{16}N_2O_2S$	65.36	5.16	8.97
<b>VIIa</b>	85	211–213 <sup>a</sup> (BuOH)	54.69	3.38	7.87	$C_{16}H_{12}Cl_2N_2OS$	54.71	3.44	7.98
<b>VIIb</b>	70	113–114 (MeOH)	70.89	6.46	8.19	$C_{20}H_{22}N_2OS$	70.97	6.55	8.28
<b>VIIc</b>	74	195–197 (PrOH)	37.61	4.88	7.74	$C_{22}H_{18}N_2OS$	37.72	5.06	7.81
<b>VIIId</b>	82	230–232 <sup>b</sup> (AcOH)	50.99	3.65	11.08	$C_{16}H_{14}BrN_3OS$	51.07	3.75	11.17
<b>VIIe</b>	76	173–175 (AcOH)	69.57	5.70	8.94	$C_{18}H_{18}N_2OS$	69.65	5.85	9.02
<b>VIIIf</b>	71	180–182 (AcOH)	53.07	3.50	7.60	$C_{16}H_{13}BrN_2OS$	53.20	3.63	7.75
<b>VIIg</b>	80	191–192 (AcOH)	53.14	3.49	7.62	$C_{16}H_{13}BrN_2OS$	53.20	3.63	7.75
<b>VIIh</b>	72	113–115 (MeOH)	65.25	5.07	8.81	$C_{17}H_{16}N_2O_2S$	65.36	5.16	8.97

<sup>a</sup> Sublimated at 170°C. <sup>b</sup> Sublimated at 200°C.

a three-component condensation of acetaldehyde **II**, cyanothioacetamide **III**, and 1-(prop-1-en-2-yl)pyrrolidine **IV** in anhydrous ethanol at 20°C. The probable reaction scheme included the Knöevenagel alkene **A** formation that further alkylated the enamine **IV** via the Stork reaction [11] to give the adduct **B**. The latter underwent intramolecular transamination [12] and oxidation (apparently with air oxygen) yielding the nicotinonitrile **I** (Scheme 1).

Alkylation of compound **I** with  $\alpha$ -haloketones **Va–Vh** in an alkaline DMF solution proceeded regioselectively at the sulfur atom to form the corresponding thioethers **VIa–VIh**. The latter underwent intramolecular cyclization to give the substituted 3-amino-4,6-dimethylthieno[2,3-*b*]pyridines **VIIa–VIIh**, promising components of antidotes of hormonal action herbicide [13, 14], sunflower growth promoters [15] and inhibitors of the *C*-terminal hydrolase L1 (UCH-L1) [16].

The determined physico-chemical and spectral parameters confirmed the structures of all the

synthesized compounds (Tables 1 and 2). Their IR spectra contained characteristic absorption bands of stretching vibrations at 2218–2225 ( $C\equiv N$ ) and 1667–1714  $cm^{-1}$  ( $C=O$ ). The  $^1H$  NMR spectra contained the signals of methyl groups (2.14–2.79 ppm), pyridine H<sup>5</sup> proton (7.00–7.09 ppm), and  $SCH_2$  fragment (for compound **VI**, 4.15–4.84 ppm).

$^1H$  NMR spectra of compounds **VIIa–VIIh** contained the signal of amino group (6.81–8.08 ppm) instead of that of  $SCH_2$  protons, thus confirming the occurred intramolecular cyclization [17–19]. In addition, absorption bands at 3180–3385 and 1638–1648  $cm^{-1}$  appeared in the IR spectra of thienopyridines **VII**, assigned to stretching and deformation vibrations of the amino group; no absorption bands of the nitrile group stretching was registered.

In summary, the three-component condensation of acetaldehyde, cyanothioacetamide, and 1-(prop-1-en-2-yl)pyrrolidine yields 4,6-dimethyl-2-thioxo-1,2-dihydronicotinonitrile that can be further converted into potentially biologically active substituted 2-alkyl-

**Table 2.** IR,  $^1\text{H}$  NMR and gas chromatography–mass spectrometry data for 2-alkylsulfanyl-4,6-dimethylnicotinonitriles **VIa–VIh** and 3-amino-4,6-dimethyl-2-Z-thieno[2,3-*b*]pyridines **VIIa–VIIh**

Comp. no.	$\nu$ , $\text{cm}^{-1}$		$m/z$ $[M + 1]^+$	$\delta$ , ppm ( $J$ , Hz)		
	C≡N or $\text{NH}_2$	C=O, $\delta(\text{NH}_2)$		Me	$\text{SCH}_2$ or $\text{NH}_2, \text{H}^5$	other signals
<b>VIa</b>	2219	1714	352	2.14, 2.38	4.78, 7.03	7.83 d (1H, $\text{H}_{\text{Ar}}$ , $J$ 7.5), 8.01 d (1H, $\text{H}_{\text{Ar}}$ , $J$ 7.5), 8.29 s (1H, $\text{H}_{\text{Ar}}$ )
<b>VIb</b>	2222	1703	339	2.17, 2.38	4.80, 7.03	0.84 t (3H, Me, $J$ 6.5), 1.11–1.72 m (4H, 2 $\text{SH}_2$ ), 2.67 t (2H, $\text{SH}_2$ , $J$ 6.4), 7.36 d (2H, $\text{H}_{\text{Ar}}$ , $J$ 8.2), 7.98 d (2H, $\text{H}_{\text{Ar}}$ , $J$ 8.2)
<b>VIc</b>	2220	1710	359	2.18, 2.37	4.84, 7.01	7.45 d (2H, $\text{H}_{\text{Ar}}$ , $J$ 6.6), 7.51 t (1H, $\text{H}_{\text{Ar}}$ , $J$ 6.9), 7.72 t (2H, $\text{H}_{\text{Ar}}$ , $J$ 7.0), 7.84 d (2H, $\text{H}_{\text{Ar}}$ , $J$ 8.4), 8.14 d (2H, $\text{H}_{\text{Ar}}$ , $J$ 8.4)
<b>VID</b>	2224	1699	377	2.39 (2Me)	4.15, 7.07	7.42 d (2H, $\text{H}_{\text{Ar}}$ , $J$ 7.1), 7.62 d (2H, $\text{H}_{\text{Ar}}$ , $J$ 7.1), 10.36 br. s (1H, NH)
<b>VIe</b>	2223	1705	295	2.22, 2.35	4.63, 7.02	2.15 s (3H, Me), 2.18 s (3H, Me), 6.88–7.00 m (2H, $\text{H}_{\text{Ar}}$ ), 7.83 d (1H, $\text{H}_{\text{Ar}}$ , $J$ 7.6)
<b>VIIf</b>	2220	1702	362	2.16, 2.38	4.78, 7.03	7.75 d (2H, $\text{H}_{\text{Ar}}$ , $J$ 7.5), 8.00 d (2H, $\text{H}_{\text{Ar}}$ , $J$ 7.5)
<b>VIg</b>	2218	1714	362	2.32, 2.39	4.15, 7.07	7.32 s (5H, Ph)
<b>VIh</b>	2225	1699	313	2.16, 2.39	4.82, 7.04	7.43–7.72 m (2H, $\text{H}_{\text{Ar}}$ ), 8.04 d (2H, $\text{H}_{\text{Ar}}$ , $J$ 8.1)
<b>VIIa</b>	3322, 3211, 3190	1695, 1648	352	2.56, 2.79	8.06, 7.05	7.68–7.79 m (2H, $\text{H}_{\text{Ar}}$ ), 7.89 s (1H, $\text{H}_{\text{Ar}}$ )
<b>VIIb</b>	3316, 3284, 3202	1699, 1645	339	2.52, 2.76	8.00, 7.09	0.91 t (3H, Me, $J$ 6.4), 1.12–1.81 m (4H, 2 $\text{SH}_2$ ), 7.34 d (2H, $\text{H}_{\text{Ar}}$ , $J$ 8.3), 7.69 d (2H, $\text{H}_{\text{Ar}}$ , $J$ 8.3)
<b>VIIc</b>	3315, 3294, 3202	1695, 1646	359	2.51, 2.77	8.08, 7.09	7.45 d (2H, $\text{H}_{\text{Ar}}$ , $J$ 8.4), 7.72 d (2H, $\text{H}_{\text{Ar}}$ , $J$ 8.4), 7.84 s (5H, Ph)
<b>VIId</b>	3300, 3280, 3195	1667, 1642	377	2.52, 2.73	7.00, 7.05	7.48 d (2H, $\text{H}_{\text{Ar}}$ , $J$ 7.8), 7.66 d (2H, $\text{H}_{\text{Ar}}$ , $J$ 7.8), 9.52 br. s (1H, NH)
<b>VIIe</b>	3385, 3309, 3214	1708, 1647	295	2.54, 2.72	7.20, 7.07	2.16 s (3H, Me), 2.22 s (3H, Me), 6.71–6.84 m (2H, $\text{H}_{\text{Ar}}$ ), 7.88 d (1H, $\text{H}_{\text{Ar}}$ , $J$ 7.7)
<b>VIIf</b>	3322, 3277, 3180	1690, 1638	362	2.52, 2.75	8.08, 7.09	7.65 d (2H, $\text{H}_{\text{Ar}}$ , $J$ 7.2), 7.78 d (2H, $\text{H}_{\text{Ar}}$ , $J$ 7.2)
<b>VIIg</b>	3319, 3295, 3200	1711, 1640	362	2.48, 2.70	6.81, 7.01	5.30 s (2H, $\text{OSH}_2$ ), 7.40 s (5H, Ph)
<b>VIIh</b>	3333, 3298, 3213	1693, 1646	313	2.44, 2.73	8.05, 7.04	7.45–7.61 m (2H, $\text{H}_{\text{Ar}}$ ), 7.68–7.82 m (2H, $\text{H}_{\text{Ar}}$ )

sulfanyl-4,6-dimethylnicotinonitriles and thieno[2,3-*b*]pyridines.

## EXPERIMENTAL

IR spectra (KBr pellets) were recorded with an UR-20 spectrometer.  $^1\text{H}$  NMR spectra ( $\text{DMSO}-d_6$ ) were registered with a Bruker AM-300 instrument

(300.13 MHz) relative to internal TMS reference. Elemental analysis was performed with a EuroVector EA-3000 analyzer. Gas chromatography–mass spectra ( $\text{CF}_3\text{COOH}$  solution, CI) were obtained using a Hewlett-Packard 5890/5972 Chrommass GC/MS instrument equipped with an HP5-MS column. Melting points were determined with a Kofler bench. The reaction progress was monitored by TLC using Silufol

UV-254 plates, eluting with acetone–hexane (3 : 5) mixture and developing with iodine vapor and UV irradiation.

**4,6-Dimethyl-2-thioxo-1,2-dihydropyridonitrile (I).** 0.6 mL (10 mmol) of freshly distilled anhydrous acetaldehyde and 2–3 drops of triethylamine were added at 20°C to a stirred suspension of 1.0 g (10 mmol) cyanothioacetamide in 20 mL of anhydrous ethanol. The mixture was stirred during 15 min until complete homogenization. 1.1 g (10 mmol) of enamine IV was then added. The reaction mixture was stirred during 1 h and then incubated during 24 h. Next, the mixture was diluted with 10% hydrochloric acid to pH 5 and incubated during 5 h; the formed precipitate was filtered off and washed with ethanol and hexane. Yield 1.2 g (72%), yellow crystals, mp 263–265°C (EtOH) (mp 264°C [8]).

**2-Alkylsulfanyl-4,6-dimethylpyridonitriles (VIa–VIh) (general procedure).** 5.6 mL (10 mmol) of 10% aqueous KOH and the corresponding  $\alpha$ -haloketone Va–Vh were added to a stirred solution of 1.7 g (10 mmol) of pyridonitrile I in 10 mL of DMF; the mixture was stirred during 2 h and then diluted with an equal volume of water. The resulting precipitate was filtered off and washed with water, ethanol, and hexane. Parameters of the products VIa–VIh are given in Tables 1 and 2.

**3-Amino-4,6-dimethyl-2-Z-thieno[2,3-*b*]pyridines (VIIa–VIIh) (general procedure).** 5.6 mL (10 mmol) of 10% aqueous KOH solution was added to a stirred solution of 10 mmol of the appropriate sulfide VI in 15 mL of DMF; the mixture was stirred during 4 h and then diluted with an equal volume of water. The resulting precipitate was filtered off and washed with water, ethanol, and hexane. Parameters of the products VIIa–VIIh are given in Tables 1 and 2.

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